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### Microplastics in drinking water

Eerkes-Medrano, Dafne; Leslie, Heather A.; Quinn, Brian

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1 **Microplastics in drinking water: A review and assessment of an emerging**  
2 **concern**

3 Dafne Eerkes-Medrano, Heather A. Leslie, Brian Quinn

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6

7 Affiliations:

8 DEM

9 Aberdeen, Scotland, United Kingdom, [d.eerkes.medrano@gmail.com](mailto:d.eerkes.medrano@gmail.com)

10 HAL

11 Department of Environment and Health, Vrije Universiteit, De Boelelaan 1085, 1081 HV Amsterdam,  
12 the Netherlands, [heather.leslie@vu.nl](mailto:heather.leslie@vu.nl)

13 BQ

14 School of Health and Life Sciences, University of the West of Scotland, Paisley, PA1 2BE Scotland,  
15 United Kingdom, [Brian.Quinn@uws.ac.uk](mailto:Brian.Quinn@uws.ac.uk)

16

17 **Abstract**

18 The first media reports of microplastics in drinking water appeared in 2017 and were followed by  
19 several scientific publications in 2018. Three important areas to consider on the subject of  
20 microplastics (MP) in drinking water (DW) are: 1) what is the evidence of MP in DW; 2) how do MP  
21 enter DW; 3) what are the toxicological implications for humans? We review these issues by  
22 presenting the published evidence of MP in tap water, bottled water and at intake and outflow of  
23 DW treatment plants; discuss the potential routes by which MP reach these destinations; address  
24 the available evidence of potential impacts on humans of MP via DW and provide a preliminary  
25 human exposure assessment; and suggest future directions for research and approaches to address  
26 emerging concerns.

27

28 **Introduction**

29 Microplastics (MP) are now ubiquitous in the environment [1-7] with concerns for MP interactions  
30 with humans [8-10]. Humans encounter MP via foods [11-15], indoor and outdoor air [5,16] and via  
31 drinking water (DW) [15,17-20]. This paper examines the issue of MP in DW and reviews the

32 available literature to address the questions: what MP concentrations have been reported; what do  
33 we know about how MP enter the DW supply; and what are the potential implications for human  
34 health?

35

## 36 **1. Reports of microplastics in drinking water**

37 In 2017 an investigation by Orb Media first reported on plastic particles in tap water  
38 ([https://orbmedia.org/stories/invisibles\\_plastics?](https://orbmedia.org/stories/invisibles_plastics?) accessed 2/11/2018). Despite lack of peer-review  
39 the report received wide media attention, and the work was later published in PLoS ONE [15]. This  
40 work presented findings on synthetic/plastic particles in tap water from fourteen countries  
41 (Supplementary Table 1). The extraction and detection methods of Liebezeit and Liebezeit [11] were  
42 used for particle categorization; unidentified particles - termed 'anthropogenic debris' - occurred in  
43 81% of the 159 globally sourced samples. The term 'anthropogenic debris' was applied because  
44 particle composition was not tested with infrared spectroscopy. Samples collected on a cellulose  
45 filter were stained with Rose Bengal, and non-stained materials counted under a dissecting  
46 microscope. The highest and lowest mean concentrations occurred in tap water samples from the  
47 US ( $9.24 \pm 11.8$  particles/L) and Germany ( $0.91 \pm 1.29$  particles/L). Developed nations had higher  
48 average particle densities than less developed nations ( $p < 0.05$ ). Most particles were fibres (98%)  
49 with size range 0.10-5.00 mm.

50 MP have conclusively been identified in bottled water [17,18] and in samples collected at drinking  
51 water treatment plants (DWTP) [19,20] (Supplementary Table 1). Two studies investigating the  
52 presence of MP in bottled mineral water detected size ranges unattainable by previous methodology  
53 (micro-Fourier Transform Infrared Spectroscopy,  $\mu$ -FT-IR). Their use of micro-Raman spectroscopy  
54 conclusively identified polymers down to 5  $\mu$ m [17] and 1  $\mu$ m size [18]. Both studies tested DW from  
55 multiple types of packaging (e.g. glass, single use plastic bottles, returnable plastic bottles and  
56 beverage cartons) and found MP in DW from all container types, with highest average particle  
57 counts in samples from reusable plastic bottles. The respective counts were  $118 \pm 88$  particles/L [17]  
58 and  $4889 \pm 5432$  particles/L (discounting an outlying particle count from a glass bottle [18]). Small  
59 particle size fractions dominated in both studies: ~80% of particles 5-20  $\mu$ m [17] and over 90% of  
60 particles  $< 5 \mu$ m [18].

61 In studies of raw and treated water from DWTP [19,20], MP have been detected at varying  
62 concentrations. One study sampled water from DWTP in three different urban areas of the Czech  
63 Republic [19]. This investigation applied scanning electron microscopy (SEM) analysis for particle

64 counts; both micro-Raman spectroscopy and  $\mu$ -FT-IR were used for identification of particles with  
65 size of 1-10  $\mu\text{m}$  and  $>10 \mu\text{m}$  respectively in processed sample volumes ranging from a total of 9 L to  
66 27 L. MP concentrations ranged from 1383-4464 particles/L in raw water to 243-684 particles/L in  
67 treated water; the smallest particle size fractions (ranges 1-5  $\mu\text{m}$  and 5-10  $\mu\text{m}$ ) dominated in both  
68 raw and treated water samples, with 95% of particles smaller than 10  $\mu\text{m}$  [19] (Supplementary Table  
69 1). These concentrations contrast with analyses of raw and treated water from five German DWTP,  
70 which found a highest concentration of 7 particles/ $\text{m}^3$  (size range 50-150  $\mu\text{m}$ ) in the raw water of  
71 one DWTP [20] (Supplementary Table 1). The sample volumes processed in this study ranged from  
72 300-1000 L for raw water and 1200-2500 L for treated water and particles as small as 20  $\mu\text{m}$  were  
73 identified with  $\mu$ -FT-IR. Water at household water taps and from three wells was also sampled, but  
74 MP concentrations did not exceed 4 particles/ $\text{m}^3$  [20]. Scrutiny of the notable variance in MP  
75 concentrations between these two studies may consider the size of particles counted, as well as the  
76 sources for water being measured: the Czech Republic study examined DWTP drawing from surface  
77 sources (two reservoirs and a river), while the German study examined DWTP drawing exclusively  
78 from groundwater sources.

79 Reports of MP 1  $\mu\text{m}$  in size [18,19] are of particular relevance due to conjectures that smaller  
80 particles ( $<1.5 \mu\text{m}$ ) may more easily cross gut epithelial membranes [21]. The presence in reusable  
81 polyethylene terephthalate (PET) bottles of the additive Tris(2,4-di-tert-butylphenyl)phosphite  
82 (average particle counts of  $708 \pm 1024 /\text{L}$ ) could indicate leaching from the bottle [18]. In the bottled  
83 water studies, PET was among the most dominant polymer types detected, which may suggest  
84 degradation of packaging material [17,18]. Findings of synthetic particles in glass bottles suggests  
85 other sources of contamination besides the packaging itself. In the study of water from German  
86 DWTP, particles identified with  $\mu$ -FTIR analysis were composed of polyester (PEST), polyvinyl  
87 chloride (PVC), polyethylene (PE), polyamide (PA) and epoxy resin. The authors indicated that such  
88 composition may come from abrasion of plastic materials used in the purification and transport of  
89 DW (e.g. pipes are often PVC or PE with PA fittings) [20]. The polyethylene and polypropylene  
90 polymers appearing in water sampled at the DWTP in the Czech Republic were suggested to come  
91 from widespread usage of plastic materials while the presence of polyacrylamide in treated water  
92 was speculated to come from a coagulant [19].

93 The methods applied by Kosuth et al. [15] were based on a previous study reporting MP in German  
94 beer [11]. This investigation acknowledged that though non-stained material was referred to as MP,  
95 only spectroscopy analysis can provide conclusive proof of MP composition. This study stimulated  
96 discussion of methodology to detect MP in beverages and was followed by two further

97 investigations of MP presence in beer. One of these studies cautioned that the staining agent (rose  
98 Bengal) may false-negatively exclude some synthetic compounds or false-positively include non-  
99 plastic compounds [22]. The second study [23] built on initial methods by applying Raman  
100 microspectroscopy (RM) to accurately distinguish between synthetic and cellulose fibres in beer and  
101 bottled mineral water (Supplementary Table 1). This investigation indicated the need for further  
102 qualitative and quantitative analysis of fibres in beverages [23].

103

## 104 **2. Drinking water sources and pathways for contamination by microplastics**

105 The extent of MP presence in DW is largely unknown – only since 2018 have peer reviewed  
106 publications covered the topic. Assessment of the spread of MP in DW will require consideration of  
107 the DW pathway (Figure 1) and the vectors for MP (Figure 2) into these DW sources.

108 Water for human consumption comes from various freshwater sources (Figure 1) which are subject  
109 to exposure to MP entering the environment through various routes (Figure 2). MP are widely  
110 reported in surface waters (rivers, lakes and reservoirs), and are commonly reported in water bodies  
111 near urban and/or populated areas [24-26] as well as remote areas [27,28]. MP may enter the DW  
112 supply from any of these water sources, as found in the Czech Republic study [19] measuring MP in  
113 raw water drawn by DWTP from reservoirs and a river. As outlined in Figure 2, MP are suggested to  
114 enter aquatic environments by spills from industrial activity [29], environmental degradation of  
115 discarded plastic items [30], washing machine effluents carrying synthetic fibres [31], effluents  
116 carrying MP found in cosmetics [32] and from the physical wear of plastic items in use [33]. The  
117 presence of MP in atmospheric samples has led researchers to suggest atmospheric transport and  
118 deposition by wind or precipitation, providing a route to aquatic environments [4,27] including  
119 surface waters for DW extraction, and with ramifications for rainwater harvesting. Wastewater  
120 treatment plants can be efficient in removing large percentages of MP from the liquid fraction (e.g.  
121 examples of 72% and up to 98% removal by treatment in Netherlands and Scotland studies  
122 respectively [34,35]), but due to large loads of MP entering WWTP, the outflow of MP in treated  
123 effluent can still be significant [36,37]. The sludge fraction has been found to contain MP [34,37,38]  
124 and is commonly used for agricultural purposes, as is treated wastewater [39], providing another  
125 route of MP into surface waters [7,37]. The entry of MP from terrestrial environments into  
126 groundwater may need further study, given the contrast between MP concentrations of an EPA  
127 (Ireland) publication (up to 6500 particles/m<sup>3</sup> in untreated private well water samples) [7], and of a  
128 study of German ground water sources (concentration up to 7 particles/m<sup>3</sup>) [20].

129 Once water is collected for drinking purposes, the ensuing treatment processes - typically screening  
130 (coagulation or settlement), filtration and disinfection - likely affect the type and size of particles  
131 entering water for consumption [40]. However, water treatment processes vary and developing  
132 countries generally have low access to treated water [41]. In 2015, 159 million people still obtained  
133 their DW directly from surface water sources [42]. Reports of DW containing MP with polymer  
134 composition similar to storage packaging suggest that transportation containers may pose an entry  
135 point for degraded plastics [17,18]. Once MP are present in freshwater drawn and processed as DW,  
136 the potential impacts of ingestion by humans need to be considered.

137

### 138 **3. Potential implications for humans**

139 Data gaps in both exposure and hazard assessments preclude an adequate risk characterization of  
140 MP to humans, via DW or any other route. Our summaries are based on what is currently known.

141 We undertook a preliminary exposure assessment of MP consumption in DW based on published  
142 particle concentrations, following the approach published by the Food and Agriculture Organization  
143 of the United Nations (FAO) on Microplastics in Fisheries and Aquaculture [14]. We consider the  
144 highest reported average particle concentrations in studies of Raman-confirmed MP in treated water  
145 from a DWTP (628 particles/L [19]), of tap water (9.24 particles/L in USA samples [15]), and of  
146 Raman-confirmed MP in bottled water (4889 particles/L in reusable PET bottles [18]). These  
147 concentrations were applied to a recommended daily water consumption rate of 3 L to obtain worst  
148 case daily particle consumption rates of ~1884, ~28, and ~15,000 for the above studies respectively  
149 (Supplementary Table 2).

150 Once ingested via DW, there is the potential for exposure to micromolecules sorbed to the MP.  
151 Chemical contaminants in DW and any additives in plastic materials will partition between MP and  
152 the aqueous phase according to fugacities of the chemicals in the water phase and the plastic phases  
153 [43]. Data on highest concentrations of various environmental contaminants and additives found in  
154 MP and on total dietary intake were obtained from Lusher et al. [14] and used to calculate the ratio  
155 of contaminant intake by DWTP treated water, tap water, and bottled water over total dietary  
156 intake (Supplementary Table 2). Based on these scenarios, MP concentrations in DW would  
157 contribute a small fraction ( $1.8 \times 10^{-9}$  to  $9.9 \times 10^{-4}$ ,  $8.6 \times 10^{-6}$  to 4.6 % and  $4.2 \times 10^{-8}$  to 0.02 % for treated  
158 water, tap water and bottled water studies respectively) of the total dietary intake of environmental  
159 contaminants and additives (Supplementary Table 2). Interesting, although bottled water had the  
160 highest number of MP, their mass was considerably lower than that of tap water owing to their

161 small size, highlighting the importance of reporting MP dimensions (shape, length, width). The  
162 applicability of these values to studies of MP in DW will depend on the presence of these chemicals  
163 in the DW supply and the values used were from contaminated environmental sites, likely to be  
164 considerably higher than in DW. It should also be noted that there are other potentially significant  
165 routes of exposure such as food, inhalation or household dust [5,12,16].

166  
167 MP may elicit particle toxicity in the human body even without leaching micromolecules. MP present  
168 in the body can produce oxidative stress [44,45], which may lead to chronic inflammation and tissue  
169 damage. Analogous to the toxicities of non-plastic particles and fibres, the size and shape of the MP  
170 may influence toxicological risk [46]. A recent review investigating MP particle uptake/translocation  
171 reported on potential effects in humans, citing gastrointestinal tract absorption as a possible uptake  
172 route [8]. Uptake and translocation to secondary target organs depends on many factors (size,  
173 surface characteristics) with smaller particles translocating more efficiently. Nevertheless, 2 µm PS  
174 particles showed a low degree of translocation across the gut layer [47]. However, PE particles up to  
175 50 µm translocated from lymph nodes into the liver and spleen, resulting in inflammatory and  
176 immune responses [48,49]. Recent reports of high concentrations of MP 1 µm and MP smaller than  
177 5 µm [17-19], indicate the potential for translocation and tissue damage, demanding further  
178 investigation for confirmation.

179

#### 180 **4. Outlook**

181 Exposure and hazard assessments of MP in DW will need to be improved before the full risks to  
182 human health can be properly understood and assessed. Exposure assessments would benefit from  
183 advances in quality assurance and quality checking (QA/QC) of sampling and analysis, the  
184 development of proficiency testing schemes and certified reference materials and from further  
185 analytical capacity to accurately detect and identify ultrafine plastic particles (i.e. in the nano-size  
186 range which is most bioavailable and likely to cause particle toxicity). Characterization of the types  
187 and magnitude of MP hazards requires an understanding of absorption, distribution and elimination  
188 of these particles, the mechanisms of toxic action, the dose-response relationship and of which  
189 human populations are at risk. Hazard data coupled with reliable real-world measured MP exposure  
190 concentrations that include both mass quantity and particle size information will ultimately enable  
191 risk characterization.

192 MP in DW represents one of many leakages of plastic debris from technical cycles into biological  
193 cycles worldwide. Strong regulatory action may be imperative to address environmental  
194 contamination problems of this nature. Furthermore, public tolerance for contaminants in DW is  
195 notoriously low and “nocebo” effects - actual adverse outcomes resulting from the perception of  
196 toxicants present [50] - can be expected for MP as much as for any other emerging contaminant in  
197 DW. Since plastic recycling cannot fully address the problem of MP emissions, a shift in focus from  
198 end-of-pipe solutions towards preventative measures is widely supported [e.g. 1,10,14,29,51].  
199 Cleaner production and processes of the circular economy (e.g. reduce, redesign) can have multiple  
200 benefits over end-of-pipe solutions. Cleaner production can be more easily coupled to profitable  
201 business models and long-term feasibility [52] while raising public awareness of pertinent issues.

202

203

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207 conversation on the water cycle and Dr Kenneth Nisbet for assisting in the risk assessment  
208 calculations.

209

## 210 **Figure captions**

211

212 Figure 1. Sources of drinking water and the various transport routes taken towards human  
213 consumption

214 Figure 2. Routes of entry for microplastics leading to drinking water sources are indicated by black  
215 arrows.

216

217

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231 natural material and the remainder (33%) contained petrochemicals. The authors reported fibres  
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247 pathways of plastics and their associated chemicals to humans as well as what is known about  
248 impacts on health. They draw attention to the need for putting individual exposure pathways in  
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279 common polymer type was poly(ethylene terephthalate) (PET). The amount of microplastic varied  
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288 drinking water treatment plant and at the water meter and the water tap of a household  
289 downstream of the drinking water treatment plant. Well ground water samples were also taken.  
290 Presence of microplastic particles ranged from 0 to 7 particles/m<sup>3</sup> with an overall sample mean of  
291 0.7 particles/m<sup>3</sup>. This was the first study to sample large volumes of water (300-1000L of raw water)  
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| Volume of DW measured (number of samples)  | Location where samples were sourced   | Volumes collected per sample; water collection vessel  | Particle identification method   | Minimum and maximum values; overall mean concentration   | Size range of particles   | Type of particles  | Comments |
|--|---|--|--|--|---|--|----------|
| Water (n=159)  | Fourteen countries spanning five continents   | 500 ml<br>For collection of 156 tap water samples, tap water was run for one minute and then a 500mL HDPE bottle was filled. Three samples of tap water were obtained in water bottles and transferred to a 500mL HDPE bottle.   | Samples were run through a Whatman cellulose filter with pore size of 2.5µm. Filters were stained with 2ml of Rose Bengal and visually analysed with a dissecting microscope. Particles were subject to a durability test - those particles that did not break were identified as anthropogenic debris.    | 0 to 60.9 particles/L; 5.45 particles/L  | 0.10-5.00 mm, with average length of 0.96 mm (only fibres were measured)  | Fibres, fragments, films. Fibres were the most dominant. |          |
| Water at drinking treatment plant (P) inlet (n=6), DWTP outlet (n=5), household water (n=5) and water tap well ground water    | Five German municipalities in the area Oldenburgisch-Ostfriesischer Wasserverband. Three wells in Holdorf | 300-1000 L of raw water, 1200-2500 L of DW at DWTP outlet<br>Water was sampled directly onto 3 µm pre-cleaned stainless steel cartridge filters placed in filter housings of SAN and PP with flexible PVC hoses attached.  | Samples were collected onto 3 µm stainless steel filters, rinsed with hydrochloric acid, then with Milli-Q and ethanol and then incubated in hydrogen peroxide. The retentate was passed through a 0.2 µm aluminium oxide filter and then analysed with micro FTIR coupled to a focal plane array detector | 0 to 7 particles/m <sup>3</sup> ; 0.7 particles/m <sup>3</sup> (14 of the 24 samples had no MP detected)   | 50-150 µm   | Fragments, fibres were suspected as contamination        |          |
| Drinking water packaged in reusable bottles (single use PET bottles (n=10), reusable bottles (n=9), single glass bottle (n=1)) | Bavarian food stores  | 250 ml of the initial sample volume. Prior to opening the bottled water containers, the exteriors were washed and dried. The contents were then transferred to pre-cleaned Erlenmeyer flasks and treated with ethylene diamine tetraacetic acid tetrasodium salt (EDTA) solution depending on calcium and magnesium ion content of the | Samples were run via vacuum through an aluminium coated polycarbonate membrane filter with pore size 0.4 µm. The filtration unit was rinsed with ethanol and then ultrapure water. Particles were analysed directly on the filter surface with micro-Raman spectroscopy.                                   | 0 to 16634 particles/L (with the outlier 35436); overall mean (calculated by the present authors) was 3633±3860.96 particles/L without outlier (4627±6785 with the outlier). Particles per L in single use PET bottles (2649 ± 2857), reusable PET bottles (4889 ± 5432), glass bottles without outlier (3074 ± 2531), glass bottle with outlier (6292 | 1 µm to >10 µm<br>The most dominant particles in single use PET bottles were in the size <1.5 µm (1419±1614) and >1.5µm to <= 5 µm range (1184±1329). Particles >5 µm (45±64). Similarly in reusable PET bottles: size <1.5 µm (2298±3048) and >1.5µm to <= 5 µm range (2365±2457). Particles >5 µm (226±307). In glass bottles: size <1.5 µm (1031±1773) and >1.5µm to <= 5 µm range | NA   |          |

|   |  |   |  |   |   |  |
|---|--|---|--|---|---|--|
| <p>samples were collected in winter when plankton occurrence is minimal.</p>  | <p>by diverse water bodies and differing in water treatment technology. Numbers of inhabitants supplied by DWTP 1, 2 and 3 are 1.5 million, 60 thousand and 130 thousand</p> | <p>over a randomised number of days. At each sampling instance 1 L of raw and treated water was collected. In total 9L and 27L each of raw and treated water were analysed for FTIR and SEM respectively. The collection vessels were pre-cleaned borosilicate glass bottles.</p>   | <p>analyses for particles 1-10 µm and &gt;10µm were performed with micro-Raman and FTIR respectively. Elemental analysis of selected particles was also performed with SEM-EDX. For SEM analysis samples were passed through polytetrafluoroethylene membrane filters. For FTIR analysis samples were passed through Al<sub>2</sub>O<sub>3</sub> filters.</p>  | <p>water from DWTP. Overall mean NA but particles/L in raw water from DWTP1 (1473±34), DWTP2 (1812±35), DWTP3 (3605±497). Particles/L in treated water from DWTP1 (443±10), DWTP2 (338±76), DWTP3 (628±28).</p> | <p>in raw (40-60% of total MP) and treated (25-60% of total MP) water. Particles 5-10 µm accounted for 30-40% particles in raw and 30-50% particles in treated water. Particles smaller than 1 µm could not be analysed for material composition but up to 2181 particles/L in raw and 230 particles/L in treated water were of this size category.</p> | <p>then fibres were the most dominant MP shapes.</p> |
| <p>packaged in returnable plastic bottles, single-use plastic bottles (n=11), beverage cans (n=3), glass bottles (n=9). Depending on whether carbonic acid was added to the water was classified as “still mineral water”, “medium sparkling” and “sparkling”</p> | <p>German grocery stores</p>   | <p>The entire water volume (between 700-1500 ml) of each water bottle or beverage carton was filtered through the pre-counted filter. Following filtration of the sample the water bottle or beverage carton was rinsed with Milli-Q water and then the filter funnel was rinsed with Milli-Q water before the vacuum was turned off.</p> | <p>Samples were filtered under vacuum through a gold coated polycarbonate filter that had been pre-counted for presence of any polymer particles (these were subtracted from sample particle counts). Both the filter and the filtering apparatus were specially manufactured to ensure a smooth surface of filter membrane and tight closure of filtration device. Particles were counted and identified with µ-Raman spectroscopy.</p> | <p>2 to 241 particles/L; overall mean NA but particles per L in single-use plastic bottles (14±14), returnable plastic bottles (118±88), glass bottles (50±52), beverage cartons (11±8)</p>                     | <p>5 µm to &gt;100 µm Particles 5-10 µm accounted for the dominant size range (41%), followed by particles 10-20 µm (30% of particles) and particles 20-50 µm (22%). Particles 50-100 µm only accounted for 5% of particles and those &gt;100 µm accounted for 2% of MP particles.</p>  | <p>fragments</p>                                     |
| <p>bottled mineral water</p>  | <p>NA</p>  | <p>3L</p>   |  | <p>1</p>  | <p>NA</p>   | <p>Fibres</p>  |

ported “anthropogenic particles” as FTIR was not applied to identify particle composition

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**Supplementary Table 2.** Comparison of the estimated intake of contaminants and additives (worst case scenario) directly from microplastics in treated water (from a water treatment plant), tap water and bottled mineral water, with their total dietary intake based on data from the FAO report on Microplastics in Fisheries and Aquaculture (Lusher et al., 2017).

| Compound             | Highest concentration in MP (ng/g) | Calculated intake from treated water (pg/kg bw/day) | Calculated intake from tap water (pg/kg bw/day) | Calculated intake from bottled water (pg/kg bw/day) | Total intake from diet (pg/kg bw/day) | Ratio intake treated water MP/total dietary intake (%) | Ratio intake tap water MP/total dietary intake (%) | Ratio intake bottle water MP/total dietary intake (%) |
|----------------------|------------------------------------|---|---|---|---------------------------------------|--|--|---|
| <b>Contaminants</b>  |                                    |   |   |   |                                       |  |  |   |
| Non-dioxin like PCBs | 2970                               | 0.0026136   | 12.2364   | 0.0594  |                                       |  |  |   |
| EFSA, 2012           |                                    |   |   |   | 4300                                  | 6.08E-05   | 0.28   | 1.38E-03  |
| JECFA, 2016          |                                    |   |   |   | 1000                                  | 2.61E-04   | 1.22   | 5.94E-03  |
| PAHs                 | 44800                              | 0.039424  | 184.576   | 0.896   |                                       |  |  |   |
| ESFA, 2008           |                                    |   |   |   | 28800                                 | 1.37E-04   | 0.64   | 3.11E-03  |
| JECFA, 2006          |                                    |   |   |   | 4000                                  | 9.86E-04   | 4.61   | 0.02  |
| DDT                  | 2100                               | 0.001848  | 8.652   | 0.042   |                                       |  |  |   |
| EFSA, 2006           |                                    |   |   |   | 5000                                  | 3.70E-05   | 0.17   | 8.40E-04  |
| JECFA, 1960          |                                    |   |   |   | 100000000                             | 1.85E-09   | 8.65E-06   | 4.20E-08  |
| <b>Additives</b>     |                                    |   |   |   |                                       |  |  |   |
| Bisphenol A          | 200                                | 0.000176  | 0.824   | 0.004   |                                       |  |  |   |
| EFSA, 2015a          |                                    |   |   |   | 130000                                | 1.35E-07   | 6.34E-04   | 3.08E-06  |
| FAO/WHO, 2011        |                                    |   |   |   | 400000                                | 4.40E-08   | 2.06E-04   | 1.00E-06  |
| PBDEs                | 50                                 | 0.000044  | 0.206   | 0.001   |                                       |  |  |   |
| EFSA, 2011           |                                    |   |   |   | 700                                   | 6.29E-06   | 0.03   | 1.43E-04  |
| JECFA, 2006          |                                    |   |   |   | 185                                   | 2.38E-05   | 0.11   | 5.41E-04  |

Note: EFSA (European Food Safety Authority), JECFA (Joint (FAO/WHO) Expert Committee on Food Additives, FAO (Food and Agriculture Organization of the United Nations), WHO (World Health Organisation), PCBs (Polychlorinated biphenyls), PAHs (Polycyclic aromatic hydrocarbons), DDT (Dichlorodiphenyltrichloroethane), PBDEs (Polybrominated diphenyl ethers), NP (Nonylphenol), OP (Octylphenol).

## Method

We followed the approach published by the Food and Agriculture Organization of the United Nations (FAO) on Microplastics in Fisheries and Aquaculture (Lusher et al. 2017) to calculate the MP contribution to total dietary intake of contaminants and additives. The highest published average microplastic concentrations for treated drinking water (628 particles/L Pivokonski et al. 2018), tap water (9.24 particles/L in USA samples, Kosuth et al. 2018) and bottled water (4889 particles/L in reusable PET bottles, Oßmann et al. 2018) were applied to a recommended daily water intake of 3 L to obtain a daily particle consumption rate for drinkers of treated water (1884) tap water (27.72) and bottled water (14,667). Estimates for particle mass consumed per kg of body weight were then approximated as below.

**Treated drinking water:** In a worst case scenario, treated drinking water from a water treatment plant contained 628 particles/L (Pivokonski et al. 2018). Polyethylene terephthalate (PET) was identified as the dominant polymer and 95% were  $\leq 10 \mu\text{m}$ . Particles were described as predominantly fragments, but no dimensions were presented. For our estimates we assume sphere shaped particles  $10 \mu\text{m}$  in size. Consumption of 1884 particles/day of  $10 \mu\text{m}$  sphere shaped PET particles (density of  $1.38 \text{ g/cm}^3$ ), represent  $0.06 \mu\text{g}$  of plastic or  $0.0009 \mu\text{g/kg b.w.}$  of an average human of 70 kg.

**Tap water:** In the tap water study most particles (98.3%) were fibres with average length  $0.96 \text{ mm}$  (Kosuth et al. 2018). Although fibre width was not reported, fibres from environmental samples had a width of  $6\text{-}175 \mu\text{m}$  [45]. Assuming  $100 \mu\text{m}$  as an estimated fibre width with a density of  $1.38 \text{ g/cm}^3$  (density of PET, the most common MP polymer found in drinking water studies (Oßmann et al. 2018)), the 27.72 particles/day represent  $288 \mu\text{g}$  of plastic or  $4.12 \mu\text{g/kg b.w.}$  of an average human of 70 kg.

**Bottle water:** The bottled water study reporting highest MP counts (4889 particles/L in reusable PET bottles, Oßmann et al. 2018) found >90% of the particles were  $<5 \mu\text{m}$  and predominantly polyethylene terephthalate (PET). Particle shape was not indicated. We assume sphere shaped particles  $5 \mu\text{m}$  in size for the purpose of our estimates. Consumption of  $5 \mu\text{m}$  sphere shaped PET particles (density of  $1.38 \text{ g/cm}^3$ ), the 14,667 particles/day represent  $1.3 \mu\text{g}$  of plastic or  $0.02 \mu\text{g/kg b.w.}$  of an average human of 70 kg.

Estimates of MP contribution to total dietary intake of contaminants and additives are conservative due to the caveat of approximating mass of particles. Volumes were calculated on assumed particle characteristics, e.g. a uniform fibre width in the case of tap water particles or a sphere in the case of bottled and treated drinking water particles. However, it is likely particle shapes vary, which would lead to ranges in the estimated mass. For more accurate representations of particle mass, the approach would have to be done gravimetrically or with the help of a mass spectrometer.

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