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Optical and structural properties of gradient (Ti,Co)Ox thin-film coatings with resistive switching effect

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Abstract: In this work, the optical and structural properties of gradient (Ti,Co)Ox coatings with resistive switching effect have been outlined. They were prepared using multimagnetron sputtering and, despite the high cobalt content, they were transparent and had a high refractive index. Gradient Co-addition resulted in the receiving fine crystalline TiO₂ – anatase and Co₃O₄ forms in the amorphous surrounding. Observed resistance switching was a fully repeatable effect, and its occurrence in gradient (Ti,Co)Ox coatings was not reported earlier. The prepared gradient coatings exhibit great potential as transparent electronic devices with resistance switching effect. Such memory effects in transparent thin-film coatings open up new possibilities for the manufacturing of innovative memory elements in the future.

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1. Introduction

Optical coatings based on metal oxides have been known and used for many years, especially in optics and optoelectronics. Currently, they are very popular in transparent oxide electronics, where various types of transparent electronic components can be produced on their basis. In the last decade, there has also been a lot of interest from various research groups in structures with resistive switching (RS) effect. However, the current state of the knowledge in the field of their operation is still insufficient that limits their application in practice [1-9]. Resistive switching devices are usually made on the basis of multilayers of various transparent metal oxides [10-12], therefore reports on the application of single gradient thin films in the construction of an RS device as an active layer instead of a typical multilayer stack is a new idea that has already been presented in our previous works [3,5]. It should also be noted that the gradient layers themselves are not commonly used today, especially in the construction of optical coatings. One of the most commonly used materials in the construction of optical coatings is titanium dioxide, of which many advantages and possibilities are already known [13-15]. It is also a good matrix for other elements, including cobalt. Cobalt oxides are one of the most studied oxides due to their importance in various fields such as supercapacitors, electrochromic sensors, solar selective absorbers, and energy storage devices [16-18]. In general, cobalt oxide can appear in three well-known forms: cobalt oxide (CoO), cobaltic oxide (Co₂O₃), and cobaltous oxide (Co₃O₄). Co₃O₄ is especially interesting due to the p-type of electrical conduction and behaves like an antiferromagnetic (AF) with a Neel temperature T_N of 290 K, it has a multiple direct band gap (2.28 eV, 1.57eV) [17-20]. The advantages briefly mentioned here have not yet

47 been fully exploited. A literature review shows that materials based on (Ti,Co)Ox have not
48 been well described in the literature so far. This is due to the fact that these types of oxide
49 mixtures are usually in the form of nanotubes, nanowires, nanorods or nanoparticles, while thin
50 films are a minority [13,14,16-23]. Usually, the amount of cobalt in the titanium dioxide matrix
51 does not exceed 15 at.%, which in the case of thin films is associated with a significant
52 reduction of their transparency above that content. The classification of oxide materials based
53 on Ti and Co mixtures is difficult to reliably perform because of the dominant influence of the
54 technology of their preparation. The key factor is the influence of the manufacturing method
55 on the structure [18]. In the case of coatings, the preparation of crystalline ones usually requires
56 additional annealing [13,14]. Without heat treatment or with a low deposition temperature,
57 amorphous films can be received [15,16]. For this reason, the use of such mixed oxides in
58 modern transparent electronics, where organic substrates are often used, indicates their high
59 application potential. In this work, the optical and structural properties of gradient (Ti,Co)Ox
60 coatings with a resistive switching effect are outlined. The films were prepared using multi-
61 magnetron sputtering. The films obtained were transparent, and their structure changed from
62 rare amorphous to fine nanocrystalline with an amorphous surrounding. As a result of the
63 aforementioned high transparency, their use in optical coatings technology seems to be a chance
64 to initiate work on a new type of transparent oxide electronic elements with memory properties.

65 **2. Experimental**

66 *2.1. Preparation of the coatings*

67 Gradient (Ti,Co)Ox thin-film coatings were prepared using the multimagnetron sputtering
68 system. Ti (99.95 %) and Co (99.95 %) metal discs (28 mm diameter and 3 mm thick) were
69 simultaneously co-sputtered in a reactive oxide mode. Pre-cleaning of the surfaces of targets
70 and substrates was applied (low-pressure plasma discharge in the chamber: AC supply, 170 W,
71 1.2×10^{-1} mbar, 3 min.). The deposition processes were carried out using a mixture of Ar:O₂
72 (1:1) as a working and reactive gas. The constant flow of each gas was set during the entire
73 deposition process at 8.4 sccm and was controlled by PR4000B (MKS) mass flow controllers.
74 The films were deposited simultaneously on three types of substrates, SiO₂, Si and Ti₄Al₆V,
75 positioned in a confocal configuration with respect to sputtered targets at a distance of 160 mm.
76 The choice of different substrates results from the requirements. For magnetrons powering two
77 separate 2 kW pulsed DC power supplies from Dora Power Systems were used [23-26]. To
78 obtain a gradient concentration of Ti and Co elements versus film thickness, the magnetron
79 equipped with the Ti target was powered with constant power fixed at 400 W, whereas the
80 magnetron with the Co target was powered with a power that increased linearly from 0 W to
81 50 W. The total time of deposition was 240 min, which resulted in 370 nm of (Ti,Co)Ox thin
82 film. Furthermore, to ensure the desired powering profile of magnetron equipped with a Co
83 target, an own-made microcontroller was used [23,24]. During the deposition, no additional
84 heating or electrical biasing was applied. Pressure during deposition was kept at about $2 \cdot 10^{-2}$
85 mbar. Details of the deposition process are collected in Table 1. To allow electrical tests,
86 circular 1 mm Au pads were evaporated at the top of the (Ti,Co)Ox thin film deposited on the
87 Ti₆Al₄V substrate.

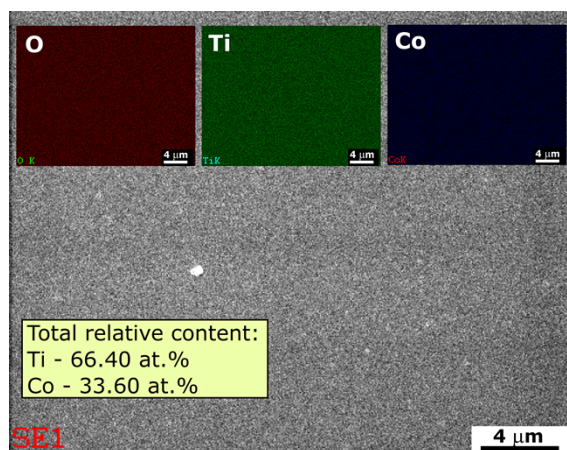
88 *2.2. Characterization of the films*

89 The surface and morphology observations at thin film cross section were performed with the
90 use of a FEI Helios NanoLab 600i SEM coupled with an EDS spectrometer. Comparison of
91 the intensity of CoK α and TiK α emission lines allowed the amount of each element to be
92 determined. Electrical properties were determined based on DC current-voltage electrical
93 measurements using a Keithley SCS4200 semiconductor characterization system (Keithley
94 Instruments LLC) and an M100 probe station (Cascade Microtech). The optical properties of
95 the gradient coatings were determined on the basis of transmission and reflection

96 characteristics registered with the aid of OceanOptics scientific grade QE65000 optical
97 spectrophotometer and a coupled deuterium-halogen lamp as a light source. In addition, also
98 an Aquila NKD-8000 spectrophotometer was used for transmission measurements. Spectra
99 were recorded for the 30° angle of incident light for S and P polarization. On their base, the
100 refractive index (n) and extinction coefficient (k), as well as optical band gap (E_g^{opt}) were
101 determined with the aid of Scout ver. 4.17 software [27]. **Characterization of structural**
102 **properties was performed on the basis of X-ray diffraction (XRD) results. The Empyrean**
103 **PIXel3D diffractometer from Panalytical was used for pattern registration. The apparatus was**
104 **operating in the grazing incidence mode (set at a 3° angle), with a Cu source. The 2theta scans**
105 **were collected in the 20° - 80° range with a step size and a time of 0.05° and 2 s, respectively.**
106 **The recorded patterns were analyzed with the aid of MDI JADE ver. 5.0 software.** For Raman
107 spectroscopy, a Thermo Scientific DXR™ Raman microscope was used. The studies were
108 performed in the range from 100 to 1000 cm^{-1} using a 455 nm blue laser diode as an excitation
109 source and a 100x objective. The power of a light source was equal to 1 mW to avoid additional
110 heating of the sample, which could result in a change of microstructure, e.g., crystallization of
111 the gradient coating. Microstructure analysis has been extended to transmission electron
112 microscopy studies. For measurements a TECNAI G2 FEG Super-Twin (200 kV) transmission
113 (TEM) and scanning electron (SEM) microscope was used. For research purposes of TEM,
114 high-quality lamellas were prepared using a focused-ion beam (FIB Quanta 3D system)
115 equipped with an Omniprobe lift-out system. The Co-gradient as a function of the film depth
116 was analyzed with a TEM-integrated X-ray energy dispersive attachment (EDS) with an
117 EDAX UTW detector.

118 3. Results and discussion

119 In Fig. 1. EDS maps of the elemental distribution of the prepared gradient (Ti,Co)Ox coating
120 are shown. As can be seen, the attached maps of O, Ti, and Co indicate a homogeneous
121 distribution of each element. The total relative content was found to be 66.40 at.% of Ti and
122 33.60 at.% of Co. It should be noted that this result is averaged over the entire volume of the
123 coating and does not contradict the fact that cobalt has a gradient distribution as a function of
124 the thin-film depth. The SEM studies also confirmed the homogeneity of the film
125 microstructure (Fig. 2). As can be seen from images of surface topography or the cross section
126 of the coating, the densely packed, columnar structure has been obtained. Near the Si substrate,
127 where the cobalt content is the lowest, the growth of the coating had a different nature than in
128 the areas closer to the surface, where its amount gradually increased.



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Fig. 1. EDS maps of the elemental distribution of the gradient (Ti,Co)Ox coating

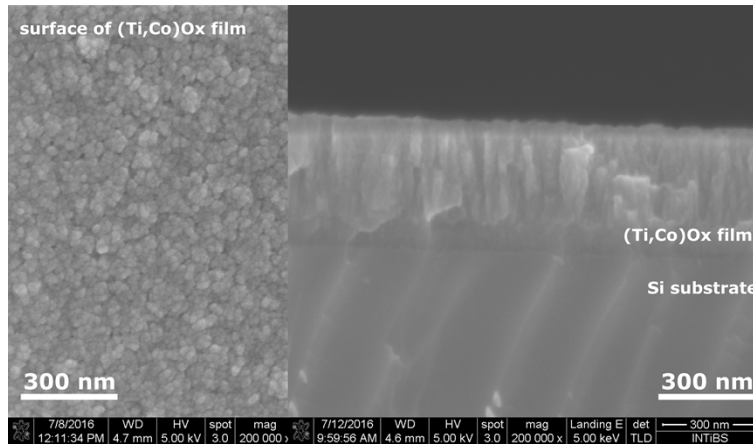
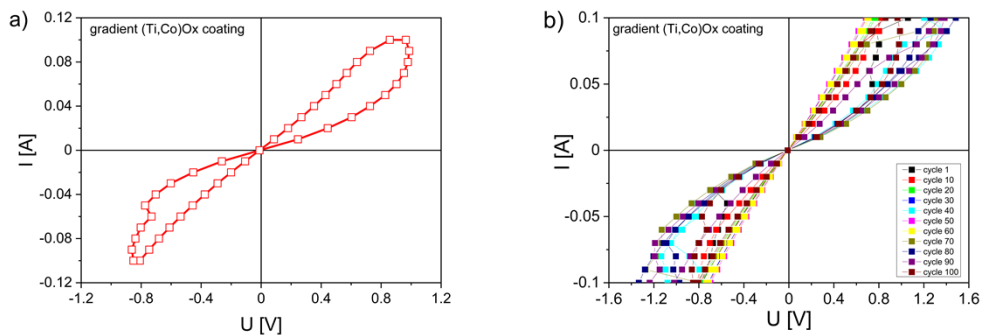


Fig. 2. SEM images of the surface and cross-section of the gradient (Ti,Co)Ox coating

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133 In Fig. 3, the I-V characteristics of the gradient coating (Ti,Co)Ox measured in the
 134 transverse configuration (Au/(Ti,Co)Ox/Ti6Al4V) are shown. Note that positive
 135 voltage/current values are due to the forward bias condition of the prepared gradient structure,
 136 whereas negative values are due to the reverse bias condition. During the measurements, the
 137 current was forced from negative to positive values and then back from the applied positive
 138 current values to the negative ones, while the voltage drop was sensed using the source-measure
 139 unit (current-driven measurement). Sweeping the current allowed us to observe the hysteresis
 140 loop (Fig. 3a). The main conclusion that can be drawn from the I-V characteristics is the
 141 presence of a bipolar resistive switching effect, which is manifested by the mentioned pinched
 142 memory loop. The occurrence of such an effect in the case of TiO₂-based coatings with the
 143 addition of cobalt has been reported so far only in a few works [3,5,18]. However, in this case,
 144 we have obtained a resistive switching effect in a gradient thin film material, which is a novelty.
 145 The repeatability of this effect is also a great advantage of the prepared gradient coating. As
 146 can be seen in Fig. 3b, the hysteresis was still present even after 100 measurement cycles. This
 147 testifies that such gradient oxide coatings based on Ti and Co are prospective for
 148 optoelectronics and gives new possibilities for preparation of various electronic devices within
 149 the framework of transparent electronics.



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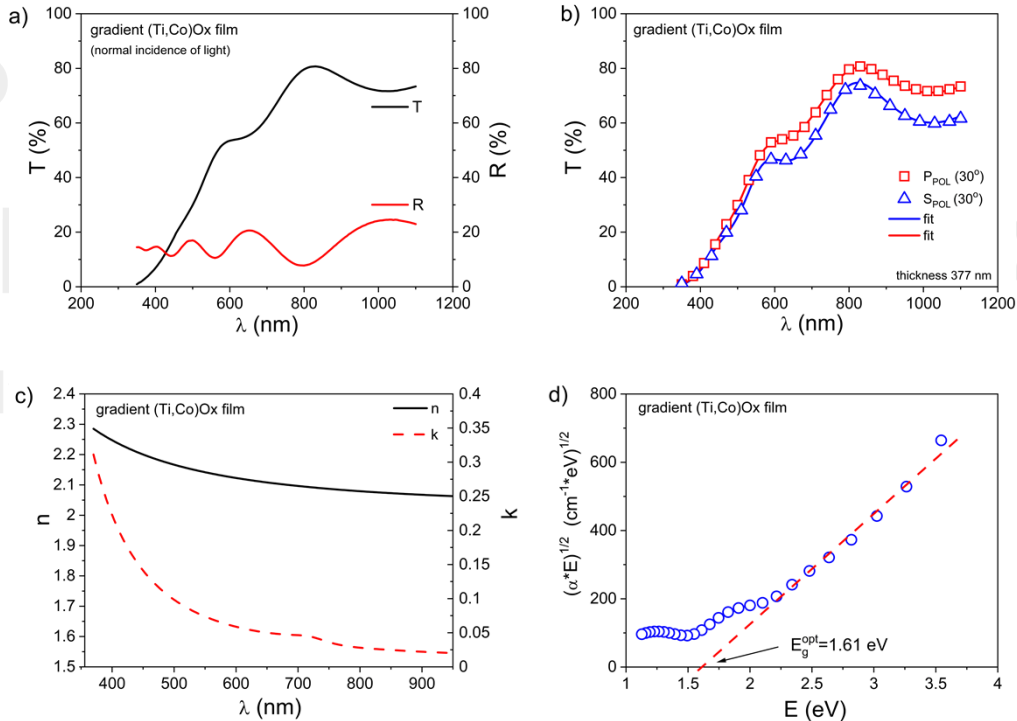
Fig. 3. I-V characteristics of the gradient (Ti,Co)Ox coating that exhibits a memristive effect (a) and its repeatability based on 100 measurement cycles (b).

153 The high attractiveness and potential possibilities offered by the application of the
 154 manufactured gradient coatings are also evidenced by the results of research on their optical
 155 properties. The transmission and reflection characteristics registered for normal incidence are
 156 presented in Fig. 4a. Also, in Fig. 4b the transmission characteristics (for S and P, measured at
 157 a 30 ° angle) are shown. These results demonstrate the high level of transparency of the

158 prepared coating. Considering the high Co content in the entire volume of the coating, the
 159 transmission coefficient above 50% for the visible light range should be considered as a very
 160 good result. Based on the characteristics of T_λ , with the help of Scout modeling [27], the
 161 characteristics of the refractive index (n) and the extinction coefficient (k) were determined
 162 (Fig. 4c). Obtained values of refractive index are typical for oxide materials based on TiO_2 ,
 163 which is widely recognized as a high- n material. For example, n_{550} was equal to 2.14. In the
 164 case of extinction coefficient, obtained values (e.g. $k_{550} = 0.07$) are 1-2 orders higher than for
 165 typical optical coatings, but it is also associated with a high cobalt content in the composition
 166 of the gradient coating. The influence of Co-additive can be observed in the band structure of
 167 the (Ti,Co)Ox film. Analysis with the aid of a Tauc plot (Fig. 4d) has shown that the optical
 168 band gap value (E_g^{opt}) was 1.61 eV. It was calculated from the $(\alpha hv)^{1/2}$ plot in the function of
 169 the photon energy (eV) [28,29]:

$$170 \quad \alpha hv = \alpha_1 (hv - E_g^{\text{opt}})^2 \quad (3.1)$$

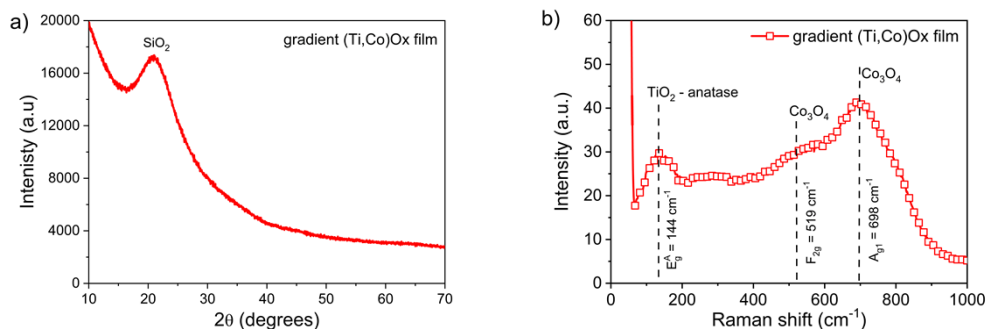
171 where: h - Planck constant, ν - radiation frequency, α_1 - constant coefficient.
 172 Obtained value is about 2 times lower than for typical TiO_2 films [13,18,25] and is a
 173 consequence of a relatively high Co-content. It should also be noted that this effect can be
 174 considered as typical for TiO_2 -based oxide materials with the addition of cobalt [18].



175
 176 Fig. 4. Characteristics of transmission and reflection registered at normal incidence of light (a), transmission for
 177 polarization of S and P at 30° angle (b); with refractive index - n and extinction coefficient - k (c), and the Tauc plot
 178 (d) of the gradient (Ti,Co)Ox coating.

179 The gradient of cobalt concentration also had an effect on the structural properties of the
 180 coating. The results of the GIXRD measurements have shown that the as-deposited (Ti,Co)Ox
 181 film was amorphous (Fig. 5a). The effect of counteracting crystallization during the deposition
 182 of the sputtered material on the substrate probably occurred. We have already observed this
 183 impact of Co-additive in the case of TiO_2 coatings with its homogeneous distribution
 184 throughout the film volume [18]. The amorphous nature of the gradient layers was also

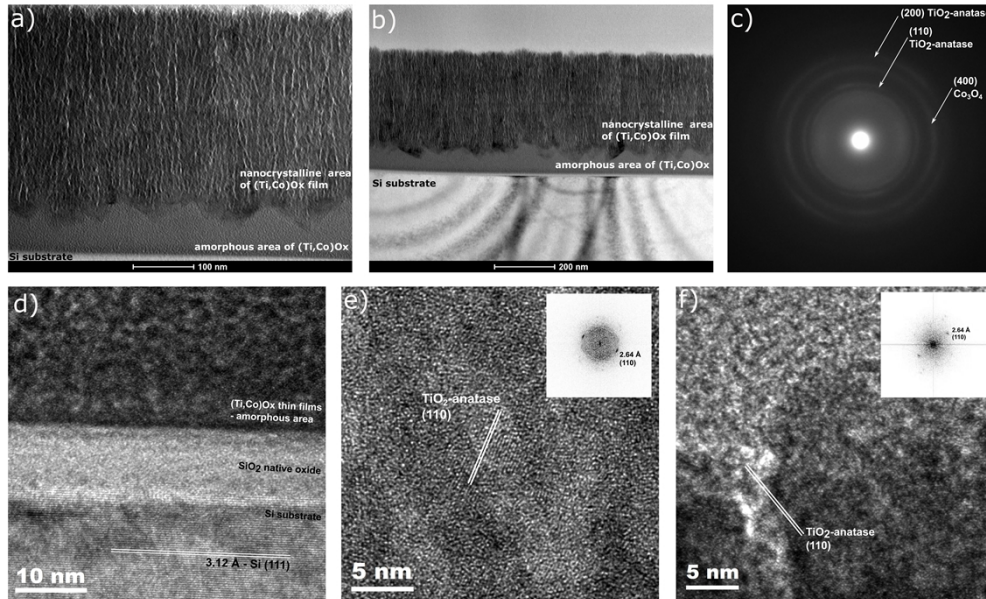
185 investigated with the aid of Raman spectroscopy (Fig. 5b). The intensity of the peaks observed
 186 in the measured spectrum is very small, suggesting that the gradient coating is rather
 187 amorphous. However, as shown, very fine crystal forms of TiO₂-anatase and Co₃O₄ are present.
 188 The presence of TiO₂-anatase is indicated by the E_g^A peak at 144 cm⁻¹ [30,31], while the
 189 presence of Co₃O₄ is suggested by the F_{2g} and A_{1g} peaks at 519 cm⁻¹ and 698 cm⁻¹, respectively
 190 [32-34]. Note that the Raman results did not give an unambiguous answer as to the amorphous
 191 or very fine crystalline structure of the (Ti,Co)Ox gradient film. Therefore, further studies were
 192 carried out using transmission electron microscopy.



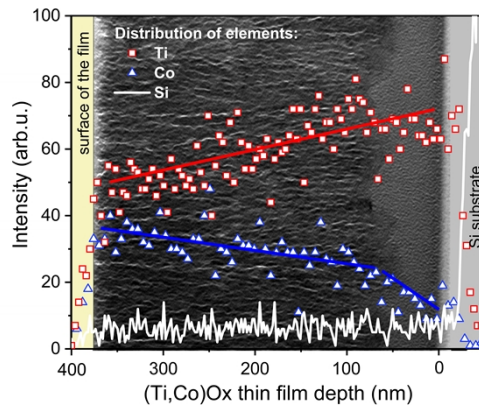
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 194 Fig. 5. GIXRD pattern (a) and Raman spectrum (b) of the gradient (Ti,Co)Ox coating as deposited on SiO₂

195 In Fig. 6 microstructure of gradient (Ti,Co)Ox coating can be observed on the cross section
 196 performed using the BFTEM, HRTEM and SAED techniques. The existence of a purely
 197 amorphous area was identified directly at the silicon substrate. It should be noted that the
 198 substrates were not heated prior to deposition; therefore, at the beginning of the nucleation the
 199 thin film remained amorphous. In turn, at a distance of approximately 50 nm from the substrate,
 200 when the amount of cobalt increased, the structure of the (Ti,Co)Ox coating was changed to
 201 fine nanocrystalline and densely packed, although much of the amorphous phase is still present
 202 (Fig. 6a,b). High-resolution images (HRTEM) were obtained to precisely determine the
 203 structure of the deposited gradient thin films and showed small crystallites with a size not
 204 exceeding 5 nm, which are located in a predominant amorphous surrounding (Fig. 6d-f). The
 205 HRTEM images with FFT (Fast Fourier Transform) diffraction patterns, taken from the
 206 crystallites free of defects, allowed for accurate evaluation of the plane separation values in
 207 three different parts of the thin films, i.e., at the substrate (Fig. 6d), in the middle (Fig. 6e)
 208 and at the surface (Fig. 6f) of the gradient coating. The HRTEM image at the substrate showed three
 209 areas of the crystalline silicon substrate, amorphous native silicon oxide, and amorphous
 210 (Ti,Co)Ox thin film. The other two images showed small nanocrystallites with marked (110)
 211 planes of TiO₂-anatase crystalline phase. The visible fringe spacings are equal to ca. 2.64 Å,
 212 which corresponds well to the separation between the (110) plane of the TiO₂-anatase phase.
 213 The selected area electron diffraction pattern (Fig. 6c) also showed rather fuzzy, broadened
 214 diffraction rings, which dominated over the rings with low intensity related to the highly
 215 nanocrystalline structure, testifying about the occurrence of the large volume of the amorphous
 216 phase in the thin film. However, three rings can be distinguished, which corresponded to the
 217 planes (200) and (110) of TiO₂-anatase and the plane (400) of the crystalline phases of Co₃O₄.
 218 This, in turn, confirmed the results obtained with the Raman spectroscopy measurements. The
 219 observed change of the microstructure character from amorphous to crystalline is combined
 220 with the increase in Co-content in the film. In Fig. 7 distribution of elements (linear scan across
 221 film thickness) with BFTEM of film cross section have been stacked together. On the basis of
 222 the linear approximation, two areas of gradient profile for Co can be distinguished. The first
 223 one from the Si substrate up to about 50 nm of layer thickness has a greater slope due to the
 224 higher sputtering rate of cobalt. In this region, as mentioned above, the layer is amorphous. In

225 the rest, where the layer already has a fine crystalline structure, a smaller slope of the gradient
 226 profile can be noticed, which corresponds to a decrease in the sputtering rate. This effect is
 227 most likely related to the poisoning of the Co-target surface with oxygen, which was used as a
 228 reactive gas. As a result, an oxide was formed on its surface, which sputtering rate was not as
 229 effective as the metallic one [35,36].



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 232 Fig. 6. Microstructure of the gradient (Ti,Co)Ox coating: BF TEM images of the cross section of the film (a,b); its SAED (c) with HRTEM images of the area near the Si substrate (d), in the middle (e) and at the top of the film (f)



233
 234
 235 Fig. 7. BFTEM cross-section image of the gradient (Ti,Co)Ox coating with distribution of elements

236 4. Conclusions

236 This paper presents studies of oxide coatings based on a mixture of titanium and cobalt with
 237 gradient cobalt concentration. The prepared structure exhibits great potential as a transparent
 238 electronic device with resistance-switching effect. Such memory effects in transparent thin-
 239 film coatings open up new possibilities for the manufacturing of innovative memory elements
 240 in the future. Coatings were prepared by multi-magnetron sputtering and, despite the high
 241 cobalt content, were transparent and had a high refractive index. Gradient Co-addition resulted

242 in the receipt of fine crystalline TiO₂ – anatase and Co₃O₄ forms in the amorphous surrounding.
243 Observed resistance switching was a fully repeatable effect, and its occurrence in gradient
244 (Ti,Co)Ox coatings was not reported earlier.

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248 **Disclosures.** The authors declare no conflicts of interest.

249 **Data availability.** Data underlying the results presented in this article are not publicly available at this time, but may
250 be obtained from the authors upon reasonable request.

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